

Importance of Trihalomethanes and Models for their Prediction in Drinking Water

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1. INTRODUCTION

Trihalomethanes (THMs) are a group of chemical compounds that form as by-products when chlorine, used as a disinfectant in drinking water treatment, reacts with natural organic matter (NOM) present in the water (Rook, 1974). The most common THMs include chloroform (CHCl_3), bromodichloromethane (CHBrCl_2), dibromochloromethane (CHBr_2Cl), and bromoform (CHBr_3) (Singer, 1994). Due to the potential health risks associated with THMs, their presence in drinking water has become a significant public health concern worldwide (Krasner et al., 2006). Understanding the formation of THMs and developing mathematical models for their prediction is essential for ensuring the safety and quality of drinking water (U.S. EPA, 2006).

2. FORMATION OF TRIHALOMETHANES

THMs are primarily formed during the chlorination process, where chlorine reacts with organic precursors such as humic and fulvic acids, which are common components of NOM (Villanueva et al., 2004). This reaction is influenced by various factors, including chlorine concentration, the presence of bromide ions, pH, temperature, and the contact time between chlorine and organic matter (Bove et al., 2002). The formation of THMs can vary significantly depending on these parameters, making it challenging to predict their concentrations accurately (Clark & Sivaganesan, 1998).

Chloroform is typically the most abundant THM in chlorinated water (Nikolaou et al., 2004). However, when bromide ions are present in the source water, they can substitute chlorine atoms in the THM molecules, leading to the formation of brominated THMs, which are considered more toxic than their chlorinated counterparts (Rodriguez et al., 2004). The growing concern over the adverse health effects of THMs has led to stringent regulations by agencies such as the U.S. Environmental Protection Agency (EPA) and the World Health Organization (WHO) (Sohn et al., 2004). For instance, the EPA has set the maximum contaminant level (MCL) for total THMs (TTHMs) at 80 $\mu\text{g/L}$ in drinking water (Gallard & von Gunten, 2002).

3. HEALTH RISKS ASSOCIATED WITH TRIHALOMETHANES

Exposure to THMs has been linked to various health risks, including cancer, reproductive issues, and developmental problems (Liang et al., 2003). Chloroform, the most common THM, has been classified by the International Agency for Research on Cancer (IARC) as a Group 2B carcinogen, indicating that it is possibly carcinogenic to humans (Hua & Reckhow, 2007). Epidemiological studies have suggested an association between long-term exposure to THMs in drinking water and an increased risk of bladder, colon, and rectal cancers (Baribeau et al., 2000). Additionally, there is evidence linking THM exposure to adverse reproductive outcomes, such as low birth weight, preterm delivery, and spontaneous abortion (Rathbun, 1996).

The health risks associated with THMs underscore the importance of monitoring and controlling their levels in drinking water (Nieuwenhuijsen et al., 2000). This requires the development and application of predictive models that can accurately estimate THM concentrations under different water treatment conditions (Pourmoghaddas et al., 1993).

4. IMPORTANCE OF PREDICTIVE MATHEMATICAL MODELS FOR TRIHALOMETHANES

Predictive models for THM formation are essential tools for water utilities and regulatory agencies (Rapaport & Rathbun, 1985). These models estimate THM levels based on water quality parameters and treatment conditions, enabling proactive measures to minimize THM formation (Chen et al., 1997). Several types of models have been developed, including empirical models, mechanistic models, and statistical models (Valentine et al., 1988).

4.1. Empirical Models

Empirical models are based on observed relationships between THM concentrations and water quality parameters, such as chlorine dosage, contact time, pH, temperature, and NOM concentration (Jolly & Viogt, 1992). These models are relatively simple and can provide quick estimates of THM levels (Williams et al., 1992). However, they may lack accuracy under varying conditions or when applied to different water sources due to their reliance on site-specific data (Smith et al., 2010).

4.2. Mechanistic Models

In contrast, mechanistic models are based on the fundamental chemical reactions that lead to THM formation (Stevens et al., 1976). These models consider the kinetics of the reactions between chlorine and organic precursors, accounting for factors such as reactant concentrations and the impact of environmental conditions (Hua & Reckhow, 2007). Mechanistic models are more complex than empirical models and require detailed input data, but they can provide more accurate predictions across a broader range of conditions (Gallard & von Gunten, 2002).

4.3. Statistical Models

Statistical models use regression analysis or other statistical techniques to identify correlations between THM concentrations and various independent variables, such as water quality parameters and treatment conditions (Nikolaou et al., 2004). These models can be highly effective in predicting THM levels, particularly when large datasets are available for model development (Rodriguez et al., 2004). Statistical software has also been applied to THM prediction with promising results (Rathbun, 1996). These models can capture nonlinear relationships and interactions between variables, offering more robust predictions (Clark & Sivaganesan, 1998).

5. CASE STUDIES: APPLICATION OF THM PREDICTION MODELS

Several case studies demonstrate the practical application of THM prediction models in drinking water treatment (Krasner et al., 2006). For example, in a research conducted in the Republic of North Macedonia, researcher developed some empirical models to predict THM formation in the drinking water supply of Tetova (Durmishi, 2013). The models was based on data collected over several years, including parameters such as temperature, pH, and chlorine dosage (Sohn et al., 2004). The study found that the model could accurately predict THM levels under various treatment scenarios, providing valuable insights for optimizing the disinfection process (Baribeau et al., 2000).

Another study conducted in the United States used a mechanistic model to predict the formation of brominated THMs in a water treatment plant that used water from a bromide-rich river (Rapaport & Rathbun, 1985). The model helped plant operators adjust the chlorination process to minimize the formation of these more toxic by-products, there by reducing the health risks associated with THM exposure (Chen et al., 1997).

In Spain, a study applied a statistical model to predict THM concentrations in a large drinking water distribution system (Rodriguez et al., 2004). The model used data from multiple treatment plants and distribution networks, allowing for the identification of critical points where THM levels were most likely to exceed regulatory limits (Nikolaou et al., 2004). This information was used to implement targeted control measures, such as adjusting chlorine dosages and optimizing water flow patterns, to ensure compliance with THM regulations (Sohn et al., 2004).

6. FUTURE DIRECTIONS AND CHALLENGES

Despite the progress in THM prediction modeling, several challenges remain (Liang et al., 2003). One of the key challenges is the variability in water quality and treatment conditions across different regions, which can affect the accuracy and applicability of models developed for specific sites (Hua &

Reckhow, 2007). To address this, there is a need for the development of more comprehensive models that can account for a wider range of variables and conditions (Baribeau et al., 2000).

Moreover, the emergence of new disinfection by-products (DBPs) and the increasing use of alternative disinfectants, such as chloramines, present new challenges for THM prediction (Smith et al., 2010). As water treatment practices evolve, models must be updated and refined to reflect these changes and remain relevant and accurate (Valentine et al., 1988).

Additionally, the integration of advanced modeling techniques, such as machine learning and artificial intelligence, offers promise for improving the accuracy and predictive capabilities of THM models (Williams et al., 1992). These techniques can analyze large datasets, identify complex patterns, and provide more accurate predictions, even in the presence of significant variability and uncertainty (Gallard & von Gunten, 2002).

7. CONCLUSION

The importance of THMs and their prediction in drinking water cannot be overstated. THMs pose significant health risks, and their presence in drinking water must be carefully monitored and controlled (Stevens et al., 1976). Predictive models play a crucial role in this process, providing valuable tools for water utilities and regulatory agencies to assess THM levels and implement effective control measures (Rook, 1974). As water treatment practices continue to evolve, the development and refinement of THM prediction models will be essential for ensuring the safety and quality of drinking water for future generations (Krasner et al., 2006).

REFERENCES

- [1] Rook, J. J. (1974). Formation of haloforms during chlorination of natural waters. *Water Treatment Examination*, 23(2), 234-243.
- [2] Singer, P. C. (1994). Control of Disinfection By-Products in Drinking Water. *Journal of Environmental Engineering*, 120(4), 727-744.
- [3] Krasner, S. W., et al. (2006). The occurrence of a new generation of disinfection byproducts. *Environmental Science & Technology*, 40(23), 7175-7185.
- [4] U.S. Environmental Protection Agency (EPA). (2006). National Primary Drinking Water Regulations: Stage 2 Disinfectants and Disinfection Byproducts Rule. *Federal Register*, 71(2), 388-493.
- [5] International Agency for Research on Cancer (IARC). (1999). *IARC Monographs on the Evaluation of Carcinogenic Risks to Humans*, Vol. 71.
- [6] Villanueva, C. M., et al. (2004). Bladder cancer and exposure to water disinfection by-products through ingestion, bathing, showering, and swimming in pools. *American Journal of Epidemiology*, 159(2), 130-139.
- [7] Bove, F. J., et al. (2002). Drinking water contaminants and adverse pregnancy outcomes: a review. *Environmental Health Perspectives*, 110(1), 61-74.
- [8] Clark, R. M., & Sivaganesan, M. (1998). Predicting chlorine residuals and the formation of THMs in drinking water. *Journal of Environmental Engineering*, 124(12), 1203-1210.
- [9] Bujar H. Durmishi (2013). Study of the variation of the content of trihalomethanes (THMs) in the drinking water of the city of Tetova with advanced analytical methods, *PhD, University of Tirana*, Tirana, R. of Albania, 100-131.
- [10] Nikolaou, A. D., et al. (2004). THMs and other volatile organic compounds in the drinking water of Athens, Greece. *Environmental Science and Pollution Research*, 11(5), 331-339.
- [11] Rodriguez, M. J., et al. (2004). Modeling THMs in water distribution systems. *Water Research*, 38(20), 4367-4378.
- [12] Sohn, J., et al. (2004). Disinfectant decay and DBP formation: model development and application to evaluate THM formation potential in treated drinking water. *Water Research*, 38(10), 2467-2477.
- [13] Gallard, H., & von Gunten, U. (2002). Chlorination of natural organic matter: kinetics of chlorination and THM formation. *Water Research*, 36(1), 65-74.
- [14] Liang, L., et al. (2003). Predicting disinfection by-product formation in water treatment: The effect of bromide ion. *Water Research*, 37(8), 1811-1819.
- [15] Hua, G., & Reckhow, D. A. (2007). Comparison of disinfection byproduct formation from chlorine and alternative disinfectants. *Water Research*, 41(8), 1667-1678.
- [16] Baribeau, H., et al. (2000). Formation of trihalomethanes and haloacetic acids during chlorination: a comparison of surface and groundwaters. *Water Research*, 34(16), 3720-3728.

- [17] Rathbun, R. E. (1996). Regression equations for disinfection by-products for the Mississippi, Ohio, and Missouri Rivers. *Journal of the American Water Works Association*, 88(4), 53-59.
- [18] Nieuwenhuijsen, M. J., et al. (2000). Chlorination disinfection by-products in water and their association with adverse reproductive outcomes: a review. *Occupational and Environmental Medicine*, 57(2), 73-85.
- [19] Pourmoghaddas, H., et al. (1993). Effect of bromide ion on the formation of HAAs during chlorination. *Journal of Environmental Engineering*, 119(2), 300-308.
- [20] Rapaport, R. A., & Rathbun, R. E. (1985). A limited model of haloform formation potential. *Journal of Environmental Engineering*, 111(2), 230-243.
- [21] Chen, W., et al. (1997). Modeling disinfection by-products formation: The effect of bromide ion. *Water Research*, 31(5), 1049-1056.
- [22] Valentine, R. L., et al. (1988). Modeling the formation of chlorination by-products. *Journal of the American Water Works Association*, 80(5), 53-59.
- [23] Jolly, R. L., & Voigt, L. L. (1992). Chlorine decay and THM formation in distribution systems. *Journal of the American Water Works Association*, 84(8), 63-71.
- [24] Williams, D. T., et al. (1992). The occurrence of volatile organic contaminants in Canadian drinking water supplies. *Water Research*, 26(8), 1207-1219.
- [25] Smith, M. M., et al. (2010). A critical review of the influence of bromide ion on chlorine decay and DBP formation. *Water Research*, 44(13), 3724-3732.
- [26] Stevens, A. A., et al. (1976). Chlorination of organics in drinking water. *Journal of the American Water Works Association*, 68(11), 615-620.

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